

PATENT

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

IN THE APPLICATION OF:

Jon L. Howell et al

CASE NO.: CH2832 US DIV

APPLICATION NO.: - - -

PRIOR GROUP ART UNIT: 1621

FILED: HEREWITH

PRIOR EXAMINER: ELVIS PRICE

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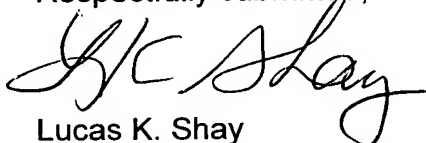
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Respectfully submitted,



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Dated: July 25, 2003

Enclosures

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Substitute for form 1449A/PTO INFORMATION DISCLOSURE STATEMENT BY APPLICANT <i>(use as many sheets as necessary)</i>				Complete if Known	
				Application Number	---
				Filing Date	July 25, 2003
				First Named Inventor	Jon L. Howell
				Group Art Unit	---
				Examiner Name	---
				Attorney Docket Number	CH2832 US DIV
Sheet	1	of	2		

[illegible][illegible]

Examiner Signature		Date Considered	
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*EXAMINER: Initial if reference considered, whether or not citation is in conformance with MPEP 609. Draw line through citation if not in conformance and not considered. Include copy of this form with next communication to applicant.

¹ Unique citation designation number. ² See attached Kinds of U.S. Patent Documents. ³ Enter Office that issued the document, by the two-letter code (WIPO Standard ST.3). ⁴ For Japanese patent documents, the indication of the year of the reign of the Emperor must precede the serial number of the patent document. ⁵ Kind of document by the appropriate symbols as indicated on the document under WIPO Standard ST. 16 if possible. ⁶ Applicant is to place a check mark here if English language Translation is attached.

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		Application Number	---
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		Examiner Name	---
Sheet 2 of 2	Attorney Docket Number	CH2832 US DIV	

OTHER PRIOR ART -- NON PATENT LITERATURE DOCUMENTS			
Examiner Initials *	Cite No. ¹	Include name of the author (in CAPITAL LETTERS), title of the article (when appropriate), title of the item (book, magazine, journal, serial, symposium, catalog, etc.), date, page(s), volume-issue number(s), publisher, city and/or country where published.	T ²
		R.N. HASZELDINE, New General Methods for the Synthesis of Fluoroiodides and Fluoroacids, Nature, Vol. 166, 192-193, 1950	
		R.N. HASZELDINE, The Reaction of Metallic Salts of Acids with Halogens. Part I. The Reaction of Metal Trifluoroacetates with Iodine, Bromine, and Chlorine, Metal Salts of Acids with Halogens. Part I, pgs 584-586, 1951	
		PASKOVICH ET AL, Simplified Method for the Preparation of Fluoroalkyl Iodides, J. Fluorine Chem. Vol 32, 833-834, 1967	
		ZHAO CGENGXUE et al, Thermal Decomposition of Some Perfluoro- and Polyfluorodiacyl Peroxides, J. Org. Chem. Vol. 47, 2009-2013, 1982	
		A. PROBAT et al, Synthesis and Chemistry of Perfluoro-2-iodo-2-Methyl-Alkanes, J. Fluorine Chem. Vol. 37, 223-245, 1987	
		A. PROBAT et al, Thermolysis and UV-Photolysis of Perfluorinated Iodo-Alkanes and Iodo-Oxaalkanes: There is a Preferred Reaction Channel, J. Fluorine Chem. Vol. 47, 163-173, 1990	
		H.K. Nair et al, Solvent effects on the reaction of Perfluoroalkyl iodides and α,ω -perfluoroalkyl diiodides with cadmium powder and dimethylcadmium, J. Fluorine Chem. Vol 60, 1-12, 1993	
		G.J. CHEN et al, Perfluoroalkylations and perfluoroalkylations. Part 2., Copper-mediated cross-coupling of secondary perfluoroalkyl iodides and aryl halides, J. fluorine Chem. Vol. 65, 59-65, 1993	
		H. FUKAYA et al, Synthesis of new nitrogen-containing perfluoroalkyl iodides, J. Fluorine Chem., Vol. 83, 117-123, 1997	

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